## Erratum to: Introducing the $\gamma$ function in quantum theory, Int. J. Quantum. Chem. 120 e26221 (2020)

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On page 9. of paper<sup>1</sup>, in Section 5., the (unnumbered) equation for the integral (11|11) contains some misprints. Its correct form is:

$$(11|11) = \frac{1}{16\pi^2 r_0^4} \iint \gamma^2(r_1 - r_0) \frac{1}{r_{12}} \gamma^2(r_2 - r_0) dV_1 dV_2$$

A more severe error is that in the next line it is wrongly stated that this expression *"is clearly divergent"*. The qualitative explanation provided there, *"the repulsion of two charged particles distributed uniformly on the surface of the same sphere is infinite."* is also false. The integral has now been evaluated correctly via expanding  $\frac{1}{r_{12}}$  in terms of spherical harmonics  $Y_{lm}$  to yield:

$$(11|11) = \frac{1}{r_0^4} \iint \delta(r_1 - r_0) \frac{1}{r_{12}} \delta(r_2 - r_0) Y_{00}^4 dV_1 dV_2$$

$$= \frac{4\pi}{r_0^4} \sum_{l,m} \frac{Y_{00}^2}{2l+1} (-1)^m \iint_0^{\infty} r_1^2 r_2^2 dr_1 dr_2 \delta(r_1 - r_0) \delta(r_2 - r_0) \frac{r_{<}^l}{r_{>}^{l+1}}$$

$$\times \iint_0^{\pi} \sin \vartheta_1 \sin \vartheta_2 d\vartheta_1 d\vartheta_2 \iint_0^{2\pi} d\varphi_1 d\varphi_2 Y_{00} Y_{lm}^*(\vartheta_1, \varphi_1) Y_{00} Y_{lm}(\vartheta_2, \varphi_2)$$

$$= \frac{1}{r_0}$$

in standard notations. Using this result for the integral (1111) in the potential energy for the He atom discussed in Sect. 5. of Ref.<sup>1</sup> corresponds to an uncorrelated (Hartree-Fock type) treatment, and results the total energy

$$E_{\rm He} = -2.25 \text{ a.u.},$$

instead of the value obtained in Ref.<sup>1</sup> in the somewhat overcorrelated "north-south" model yielding -3.06 a.u. The latter is much closer to the experimental energy of -2.90 a.u. One concludes that, apart from the erroneous statement on the integral cited above, the conclusions of Ref.<sup>1</sup> about the treatment of He remain valid.

## References

1. P. R. Surján, Int. J. Quantum Chem. 120, e26221 (2020).