

Resonance theory of catalytic action of transition-metal complexes.

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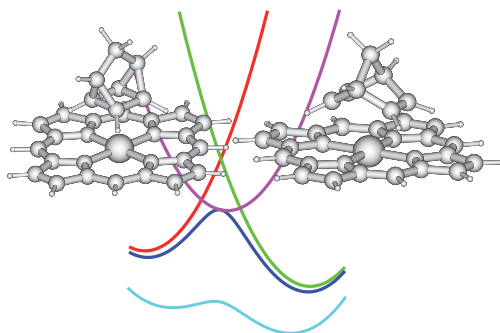
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A Google "catalysis theory" search results in dull entries like "DF Theory application to catalysis of ... by ..." rather than "theory of catalysis". We, instead, provide a general theory of catalysis of Woodward-Hoffmann restricted reactions by transition-metal complexes based on a) valence-bond ideas of chemical reactivity, b) entanglement of electronic states of reactants, and catalyst described by c) an effective Hamiltonian for the reaction center. In development of our previous works [1–3] where the potential energy surface of a catalytic reaction was expressed through the eigenstates of the free reactants Φ_r^i and catalyst Φ_d^k :

$$E = \sum_{k,i} (A_{ki})^2 (E_d^k + E_r^i) + \sum_{k,i} \sum_{k',i'} A_{ki} A_{k'i'} \langle \Phi_d^k \otimes \Phi_r^i | \mathbf{H}_{int} | \Phi_d^{k'} \otimes \Phi_r^{i'} \rangle. \quad (1)$$

we apply the program suit catal [4] based on eq. (1) to the catalytic transformation of quadricyclane to norbornadiene:



We confirm the qualitative explanation of contrasting catalytic behavior of Mn- and Co-porphyrins obtained previously [1–3] from the analysis of the spectra of local many-electron states of free catalysts and the reactant/product. A new four-state qualitative model of this catalytic isomerization is proposed [5] as based on the results of this numerical study.

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[5] A.L. Tchougréeff, A.M. Tokmachev, R. Dronkowski. *Int. J. Quant. Chem.* Accepted. DOI: 10.1002/qua.24386.